

Preferential binding of α -actinin to actin bundles

Enrico Grazi, Paola Cuneo, Ermes Magri and Christine Schwienbacher

Istituto di Chimica Biologica, Università di Ferrara, Via Borsari 46, 44100 Ferrara, Italy

Received 24 September 1992; revised version received 4 November 1992

At 37°C, the α -actinin–F-actin binding isotherm is anomalous. In 6.7% polyethylene glycol 6000, concomitantly with the formation of actin bundles, the binding isotherm becomes hyperbolic ($K_{\text{diss}} = 11.3 \mu\text{M}$). α -Actinin increases the rigidity of the networks formed by actin bundles in polyethylene glycol and by paracrystalline actin in 16 mM MgCl_2 but not by F-actin. It is proposed that in the cell α -actinin functions are mostly carried on by interaction with actin bundles.

α -Actinin function; Preferential binding; Actin bundle

1. INTRODUCTION

α -Actinin is a 200 kDa protein that crosslinks actin filaments and increases the rigidity of the actin gel. Both phenomena are quite complex and are not completely understood.

The binding isotherm of α -actinin with actin, as a function of actin concentration, is anomalous. The apparent association constant decreases with an increase in actin concentration [1,2]. This phenomenon was not recognized previously because the effect of actin concentration either was not tested [3–7] or was tested at relatively high actin concentrations [8].

The critical gelling concentration of α -actinin increases substantially with temperature [3], a phenomenon that led to questioning of the gelling activity of this protein *in vivo* [5]. It was found, however, that α -actinin is an efficient actin gelling protein, even at 37°C, provided that either the concentration of actin is low (1.2–2.4 μM) [2] or the reaction mixture is supplemented with macromolecules at a concentration equivalent to that found in the cell sap [9].

We offer evidence here that the presence of a network of actin bundles, independent of the mechanism of its formation, is a prerequisite for α -actinin functioning *in vivo*.

2. MATERIALS AND METHODS

G-actin from rabbit muscle was prepared according to Spudich and Watt [10] and further gel filtered through Sephadex G-150 [11]. α -Actinin from chicken gizzard was prepared according to Feramisco and Burridge [12]. The absorption coefficients used were $A_{280}^{1\%} = 6.2$ [13] for actin and $A_{280}^{1\%} = 9.7$ [14] for α -actinin. Molar concentrations were

calculated on the basis of a molecular mass of 42 kDa for actin [15] and of 200 kDa for α -actinin [14]. Centrifugation was performed at 37°C in a TL100 rotor of the TL100 Beckman centrifuge.

Protein was determined by the Coomassie blue method [16] as modified by Stoscheck [17].

[^3H]-N-Ethylmaleimide-labelled α -actinin was prepared and radioactivity determined as previously described [2].

The rigidity of the gels of actin was measured by the droplets method [18].

3. RESULTS

3.1. Effect of polyethylene glycol 6000 on the complex interactions of α -actinin with actin

It is known that, in the presence of polyethylene glycol (PEG) 6000, F-actin undergoes massive conversion into actin bundles [19]. At 12 μM actin the boundary between filaments and bundles ranges between 6 and 7% (w/v) PEG. Addition of 0.2 μM α -actinin to the system displaces the boundary toward a lower (4–6%) PEG concentration. This shows that α -actinin favours actin bundling (Fig. 1).

The amount of α -actinin co-sedimenting with F-actin is not influenced up to 3% PEG but increases at larger PEG concentrations. The increase is concomitant with the formation of actin bundles. Under these conditions (6.7% PEG), approximately the same amount of α -actinin is sedimented by centrifugation either at $9,900 \times g$ (actin bundles are collected) or at $366,000 \times g$ (actin bundles plus actin filaments are collected). Thus, at this PEG concentration, α -actinin is bound almost exclusively to actin bundles (Fig. 2).

The α -actinin–F-actin binding isotherm is anomalous both in the absence [1,2] and in the presence of 3% PEG (Fig. 3). Under both these conditions actin is filamentous. The binding isotherm becomes hyperbolic concomitant with the formation of actin bundles (6.7% PEG). Double reciprocal plot analysis shows that a sin-

Correspondence address: E. Grazi, Istituto di Chimica Biologica, Università di Ferrara, Via Borsari 46, 44100 Ferrara, Italy. Fax: (39) (532) 202 723.

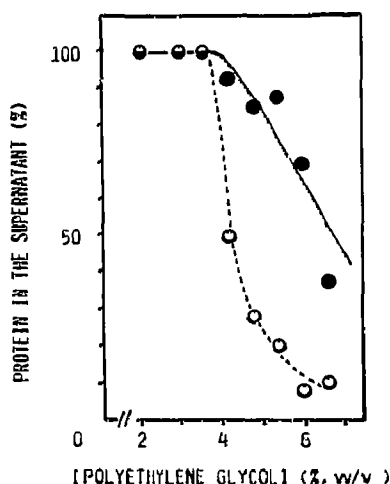


Fig. 1. In polyethylene glycol 6000 α -actinin promotes actin bundling. The mixtures contained F-actin ($12 \mu\text{M}$ as monomer), 0.5 mM ATP, 2 mM MgCl_2 , 0.1 M KCl, 1 mM dithiothreitol, 10 mM Tris-HCl, PEG 6000 (w/v) as indicated in the figure, with (○) or without (●) $0.2 \mu\text{M}$ [^3H]N-ethylmaleimide-labelled α -actinin (specific activity $12,000 \text{ dpm/nmol}$), pH 7.5. After 60 min of incubation at 37°C , the mixtures were centrifuged for 10 min at $9,900 \times g$ to sediment actin bundles, and the supernatant solutions were assayed for protein.

gle dissociation constant ($K_{\text{diss}} = 11.3 \mu\text{M}$) accounts for the binding of $0.185 \mu\text{M}$ out of the total $0.2 \mu\text{M}$ α -actinin (Fig. 3b).

3.2. The interaction of α -actinin with paracrystalline actin in 16 mM MgCl_2

In the presence of $0.2 \mu\text{M}$ α -actinin (total concentration) and of $48 \mu\text{M}$ actin, more α -actinin is bound to paracrystalline actin in 16 mM MgCl_2 ($0.091 \mu\text{M}$, Fig. 4) than to filamentous actin in 2 mM MgCl_2 plus 0.1 M KCl ($0.026 \mu\text{M}$, Fig. 3). The binding of α -actinin to

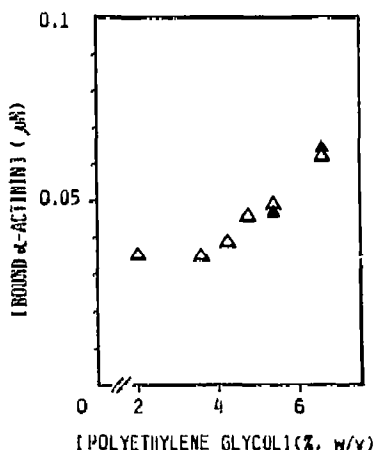


Fig. 2. Binding of α -actinin to F-actin in polyethylene glycol 6000. Experimental conditions were as described in Fig. 1. After 60 min of incubation at 37°C , the mixtures were centrifuged for 10 min either at $9,900 \times g$ (▲) or at $366,000 \times g$ (Δ). The pellets were then assayed for labelled α -actinin.

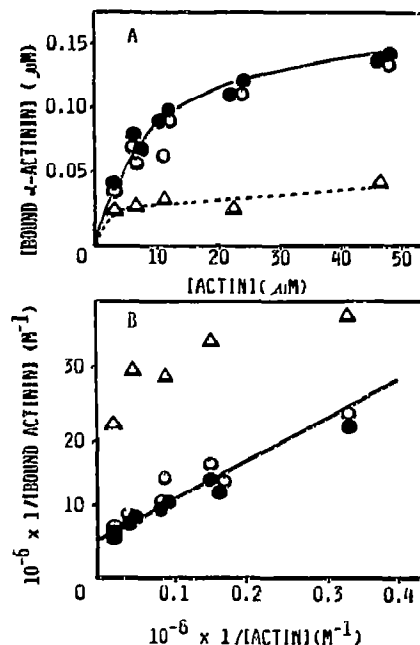


Fig. 3. Binding of α -actinin to F-actin in polyethylene glycol 6000, as a function of actin concentration. (a) Actin concentration was as indicated in the figure; PEG was either 3% (Δ) or 6.7% (○, ●). Other conditions were as described in Fig. 1. After 60 min of incubation at 37°C , the mixtures were centrifuged for 10 min either at $9,900 \times g$ (●) or at $366,000 \times g$ (○, Δ). The pellets were then assayed for labelled α -actinin. (b) Double reciprocal plot of the binding of α -actinin to F-actin.

paracrystalline actin, however, is not described by a simple hyperbolic function (Fig. 4).

The rigidity of the system ($7.1 \mu\text{M}$ actin as monomer) increases from 2.6 to 6.7 dyn/cm^2 in the transition from

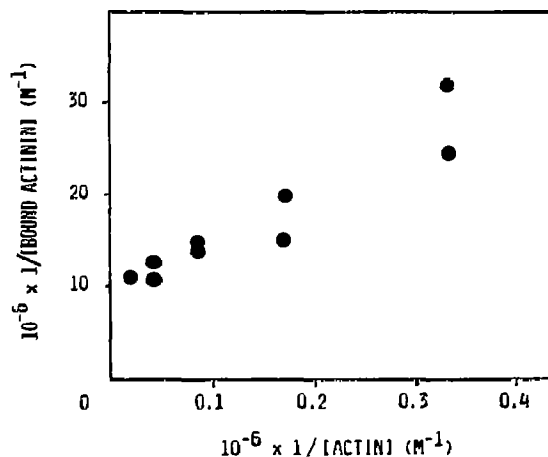


Fig. 4. Double reciprocal plot of the binding of α -actinin to actin paracrystals in 16 mM MgCl_2 . The mixtures contained F-actin as indicated in the figure, $0.2 \mu\text{M}$ [^3H]N-ethylmaleimide-labelled α -actinin (specific activity $12,000 \text{ dpm/nmol}$), 0.5 mM ATP, 16 mM MgCl_2 , 1 mM dithiothreitol, 10 mM Tris-HCl, pH 7.5. After 60 min of incubation at 37°C , the mixtures were centrifuged for 10 min at $366,000 \times g$. The pellets were then assayed for labelled α -actinin.

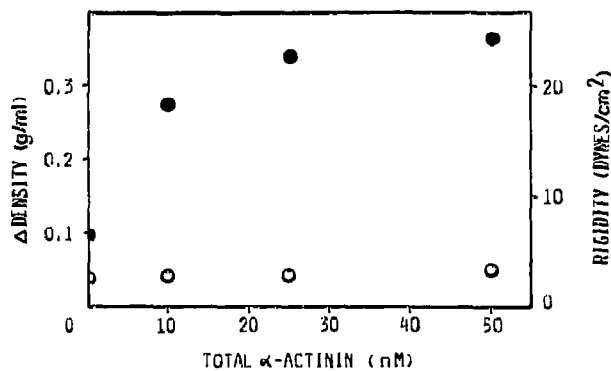


Fig. 5. Estimate of the effect of increasing α -actinin concentrations on the rigidity of the network formed by paracrystalline actin. The mixtures contained F-actin ($7.1 \mu\text{M}$ as monomer), 0.5 mM ATP, 1 mM dithiothreitol, 10 mM Tris-HCl and either 16 mM MgCl_2 (●) or 2 mM MgCl_2 plus 0.1 M KCl (○). α -Actinin concentration was as indicated in the figure. After 60 min of incubation at 37°C and pH 7.5, rigidity measurements were performed. Δ density represents the difference between the density at which the droplets remained stationary in the complete system and in the salt solution without protein.

F-actin to paracrystalline actin. The rigidity is further increased to 24.3 dyn/cm^2 when paracrystalline actin is supplemented with $0.05 \mu\text{M}$ α -actinin (Fig. 5).

4. DISCUSSION

In 6.7% PEG 6000, the binding of α -actinin to actin bundles is described by a single dissociation constant of $11.3 \mu\text{M}$. This contrasts with the anomalous behaviour displayed by F-actin and characterized by the apparent decrease of the binding constant to α -actinin, as a function of the increase of F-actin concentration.

The parallel arrays of actin filaments, formed either in 6.7% PEG 6000 or in 16 mM MgCl_2 , bind α -actinin tighter than does F-actin. This is in keeping with the observation that, in the cell, α -actinin is mostly associated with actin fibers [20,21]. It is likely that the arrays of filaments offer an ordered matrix of actin, which favours by bidentate binding of α -actinin. The crosslinking by α -actinin prevents the filaments from sliding in actin bundles. As a result, since the network of actin bundles is largely anastomosed, the rigidity of

the system is increased by α -actinin, even at 37°C . In F-actin, at least at 37°C , the monodentate binding of α -actinin prevails. This is indicated by the total lack of effect of α -actinin on the rigidity of the network formed by F-actin.

These observations support the view that, in the cell, α -actinin functions are mostly carried on by interaction with actin bundles.

Acknowledgements: This work was supported by grants of MURST 40% and 60%.

REFERENCES

- [1] Grazi, E., Trombetta, G. and Guidoboni, M. (1990) *Biochem. Int.* 21, 633–640.
- [2] Grazi, E., Trombetta, G. and Guidoboni, M. (1991) *J. Muscle Res. Cell Motil.* 12, 579–584.
- [3] Goll, D.E., Suzuki, A., Temple, J. and Holmes, G.R. (1972) *J. Mol. Biol.* 67, 469–488.
- [4] Jockusch, B.M. and Isenberg, G. (1981) *Proc. Nat. Acad. Sci. USA* 78, 3005–3009.
- [5] Bennett, J.P., Scott-Zaner, K. and Stossel, T.P. (1984) *Biochemistry* 23, 5081–5086.
- [6] Ohtaki, T., Tsukita, S., Mimura, N., Tsukita, S. and Asano, A. (1985) *Eur. J. Biochem.* 153, 609–620.
- [7] Landon, F., Gache, Y., Toulou, H. and Olomucki, A. (1985) *Eur. J. Biochem.* 153, 231–237.
- [8] Sato, M., Schwarz, W.H. and Pollard, T.D. (1987) *Nature* 325, 828–830.
- [9] Grazi, E., Trombetta, G., Magri, E. and Cuneo, P. (1990) *FEBS Lett.* 272, 149–151.
- [10] Spudich, J.A. and Watt, S. (1971) *J. Biol. Chem.* 246, 4866–4871.
- [11] McLean-Fletcher, S. and Pollard, T.D. (1980) *Biochem. Biophys. Res. Commun.* 96, 18–27.
- [12] Feramisco, J.R. and Burridge, K. (1980) *J. Biol. Chem.* 255, 1194–1199.
- [13] Gordon, D.J., Yang, Y.Z. and Korn, E.D. (1976) *J. Biol. Chem.* 251, 7474–7479.
- [14] Suzuki, A., Goll, D.E., Singh, I., Allen, R.E., Robson, R.M. and Stromer, M.H. (1976) *J. Biol. Chem.* 251, 6860–6870.
- [15] Collins, J.H. and Elzinga, M. (1975) *J. Biol. Chem.* 250, 5915–5920.
- [16] Bradford, M.M. (1976) *Anal. Biochem.* 72, 248–254.
- [17] Stoscheck, C.M. (1990) *Anal. Biochem.* 184, 111–116.
- [18] Grazi, E., Magri, E., Cuneo, P. and Cataldi, A. (1991) *FEBS Lett.* 295, 163–166.
- [19] Suzuki, A., Yamazaki, M. and Ito, T. (1989) *Biochemistry* 28, 6513–6518.
- [20] Lazarides, E. (1976) *J. Cell Biol.* 68, 202–219.
- [21] Bretscher, A. and Weber, K. (1978) *J. Cell Biol.* 79, 839–845.